

gM-Ingenieurbüro



a) gM-Ingenieurbüro, Tannenweg 2, D-35274 Kirchhain, Germany, peste@gm-ingenieurbuero.com;
b) Tour-Solaire.fr, 8 Impasse des Papillons, 34090 Montpellier, France, renaud_derichter@solar-tower.org.uk
c) School of Civil Engineering and Architecture, Wuhan University of Technology, No. 122, Luoshi Road, Wuhan, 430070, P. R. China.
tzming@whut.edu.cn

conclusions

2017: A technical variant mimicking the natural “Iron Salt Aerosols (ISA)” climate control method has been proposed¹²). 12 or more cooling effects are described, among them CH₄ and CO₂ atmospheric removal.

Does current research ignore important climate regulation principles?

Ignored: halogen activation by iron photolysis (Figure 1)

1989: Revelation of the O₃-induced atomic °Cl generation from gaseous HCl by Fe₂O₃ aerosol photolysis. This process shall contribute to the chemistry of Antarctic O₃ depletion²⁶).

2009 / 2017: Discussions about the cause of atomic $^{\circ}\text{Cl}$ and $^{\circ}\text{Br}$ in volcanic eruption plumes(27)(28)(29)(30)(31)(32).

2013 / 2017: Models about the influence of mankind induced halogen emissions on tropospheric O3 content³³⁾³⁴⁾.

2017: Discussion about the increased tropospheric halogen chemistry during glacial climate periods³⁵).

1998 / 2017: Discussion about possibilities of the generation of atomic °Cl and °Br from arctic snow and sea ice36)37)38)39)40)41)42)43).

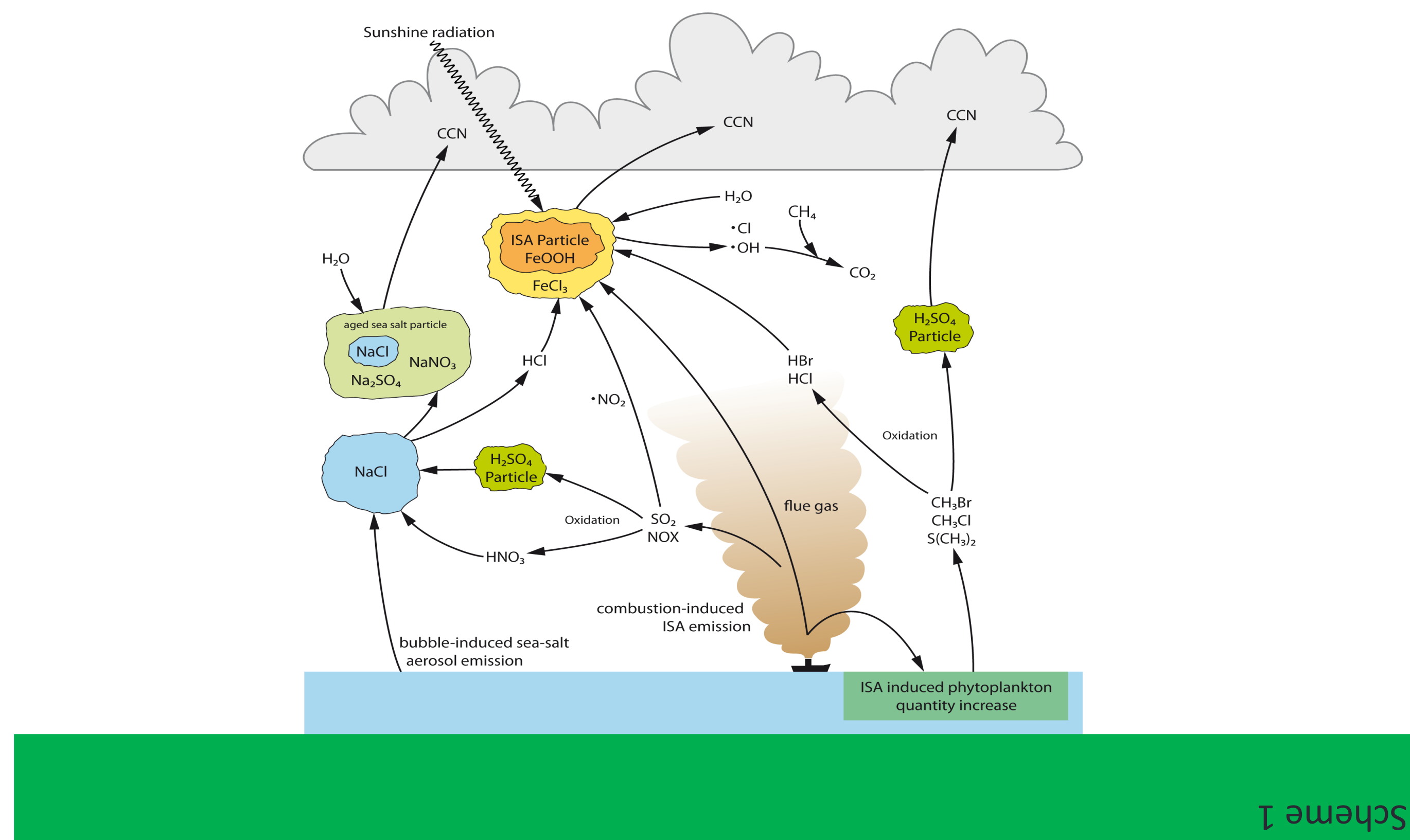
Ignored: carbon cycle direction

atmosphere → ocean → sediment / crust; not vice versa (Scheme 1):

2014: Upwelling water in the Antarctica neighboring glacial Southern Ocean should have lost lesser CO₂ to the atmosphere than recently⁴⁴).

FIGURE 1

Tropospheric CH₄ depletion induced by Fe(III) photolysis



Driving-forces directing the carbon cycle from atmosphere into ocean and from ocean into sediments and crust

Troposphere: Oxidation of CH₄, CO, COS and VOC to CO₂ and soot hydrophilization
→ → →

Continent/ocean:

- 1) CO₂ absorption
- 2) CO₂ hydrolysis to HCO₃⁻,
- 3) CO₂ and HCO₃⁻ metabolization to organic and carbonate C
- 4) sedimentation of organic and carbonate C

→ → →

Sediment/oceanic crust: Carbonate and CH₄ hydrate precipitation
→ → →

BIBLIOGRAPHY

- Weilrecht JJ, (1870): *Nachricht in der Zubereitung der gelben und weißen Eisenkieselsäuren*. St. Petersburgsches Journal, 10, 184-189
- 2) Gehlen AF, (1804). Neues allgemeines Journal der Chemie, 3, 566; aus Landgrebe DG, (1834): Über die chemischen und physiologischen Wirkungen des Lichtes. Druck und Verlag von N.G. Elwert, Marburg
- 3) Martin JH, (1990): Glacial-interglacial CO₂ change: the iron hypothesis. *Paleoceanography*, 5, 1-13
- 4) Oestle FD, (2004): Climate cooling by interaction of natural (A) or artificial (B) loess dust with tropospheric methane. *GeolEipz* 2004, Deutsche Geologische Gesellschaft, Hannover 2004, p. 344, ISBN 3-932537-06-8
- 5) Smith-Duque C, (2009): Hydrothermal alteration of upper oceanic crust formed at fast spreading rates. Thesis, University of Southampton, Faculty of Engineering, Science & Mathematics, School of Ocean & Earth sciences
- 6) Staudigel H (2003): 3.15 Hydrothermal alteration processes in the ocean crust. *Treatise on Geochemistry*, Volume 3, Elsevier, 511-535, ISBN 0-08-044338-9
- 7) Rausch S. (2012): Carbonate veins as recorders of seawater evolution, CO₂ uptake by ocean crust, and seawater-crust interaction during low temperature alteration. Thesis, Universität Bremen, Fachbereich Geowissenschaften
- 8) Fletcher BJ, Brentnall SJ, Anderson CW, Berner RA, Beerling DJ, (2008): Atmospheric carbon dioxide linked with Mesozoic early Cenozoic climate change. *Nature Geoscience*, 1, 43-48
- 9) Taylor LL, Banwart SA, Valdes PJ, Leake RL, Beerling DJ, (2012): Evaluation the effects of terrestrial ecosystems, climate and carbon dioxide on weathering over geological time: a global-scale process-based approach. *Phil. Trans. R. Soc. B*, 367, 565-582
- 10) Hartmann J, West AJ, Renforth P, Köhler P, De la Rocha, CL, et al. (2013): Enhanced chemical weathering as a geoengineering strategy to reduce atmospheric carbon dioxide, supply nutrients and mitigate ocean acidification. *Reviews of Geophysics*, 51, 113-150
- 11) Kronfeld J, Moinester M, Carmi I, Godfrey-smith D, (2015): Sequestration of inorganic carbon via Forestation. Presented at Climate Engineering Research Symposium July 2015, Berlin, Germany
- 12) Oestle FD, de Richter R, Ming T, Caillol S, (2017): Climate engineering by mimicking natural dust climate control: the iron salt method. *Earth Syst. Dynam.* 8, 1-54
- 13) Masson-Delmotte V, Stenni B, Pol K, Braconnot P, Cattani O, Falourd S, Kageyama M, Jouzel J, Landais A, Minster B, Barnola JM, Chappellaz J, Krinner G, Johnsen S, Röhlisberger R, Hansen J, Mikolajewicz U, Otto-Bliesser B, (2010): EPICA Dome C record of glacial and interglacial intensities. *Quaternary Science Reviews*, 29, 113-128
- 14) Liu Y, Shi G, Xie Y, (2013): Impact of dust aerosol on glacial-interglacial climate. *Advances in Atmospheric Sciences*, 30(6), 1725-1731
- 15) Dagsson-Waldhauserova P, Arnalds O, Olafsson H, Magnusdottir A, (2017): High proportions of sub-micron particulate matter in Icelandic dust storms in 2015. *Geophysical Research Abstracts*, 19, presented at EGU2017, 8762-1
- 16) Kubin A, Schepanski K, Heinold B, Tegen I, (2017): Sensitivity studies on glaciogenic dust mobilization in Greenland with a mesoscale model. *Geophysical Research Abstracts*, 19, presented at EGU2017, 3207
- 17) Rubasinghege G, Lentz RW, Scherer MM, Grassian VH, (2010): Simulated atmospheric processing of iron oxyhydroxide minerals at low pH: roles of particle size and acid anion in iron dissolution. *PNAS*, 107(15), 6628-6633
- 18) Lim M, Chiang K, Amal R, (2006). Photochemical synthesis of chlorine gas from iron(III) and chloride solution. *Journal of Photochemistry & Photobiology A: Chemistry*, 183(1-2), 126-132
- 19) Wittmer J, Bleicher S, Ofner J, Zetzsch C, (2015a): Iron(III)-induced activation of chloride from artificial sea-salt aerosol. *Environmental Chemistry*, 12(4), 461-475
- 20) Wittmer J, (2015b): Photochemical activation of chlorine and bromine from iron-doped saline media. Dissertation at the Biological, Chemical and Geosciences Faculty of the University of Bayreuth, Germany
- 21) Wittmer J & Zetzsch C, (2016): Photochemical activation of chlorine by iron-oxide aerosol. *Journal of Atmospheric Chemistry*, 10874, 1-18
- 22) Vione D, Maurino V, Minero C, Pelizzetti E, Harrison MAJ, Olariu R-J, Arsene C, (2006): Photochemical reactions in the tropospheric aqueous phase and on particulate matter. *Chem. Soc. Rev.*, 35, 441-453
- 23) Luna AJ, Nascimento CAO, Chivone-Vilho O, (2006): Photodecomposition of hydrogen peroxide in highly saline aqueous medium. *Brazilian Journal of Chemical Engineering*, 23(3), 341-349
- 24) Machulek A, Moraes JEF, Okano LT, Silvério CA, Quina FH, (2009): Photolysis of ferric ions in the presence of sulfate or chloride ions: implications for the photo-Fenton process. *Photochem. Photobiol.*, 85, 985-991
- 25) Machulek A, Quina FH, Gozzi F, Silva VO, Friedrich LC, Moraes JEF, (2012): Chapter 11 – Fundamental mechanistic studies of the photo-Fenton reaction for the degradation of organic pollutants. In the book titled "Organic pollutants ten years after the Stockholm Convention – Environmental and Analytical update" edited by Puzyn T and Mostrag-Szlichtyng A. ISBN 978-93537-307-917-2
- 26) Behnke W & Zetzsch C, (1989): Photochemical formation of Cl atoms from NaCl and HCl in the presence of ozone and aerosol. Our changing Atmosphere, Proceedings of the 28th Liège International Astrophysical Colloquium. Edited by P.J. Crutzen, J.-C. Gerard and R. Zander, 493-499
- 27) von Glasow R, Bobrowski N, Kern C, (2009): The effects of volcanic eruptions on atmospheric chemistry. *Chemical Geology*, 1-4, 131-142
- 28) Baker AK, Rauthe-Schöck A, Schuck TJ, et al, (2011). Investigation of chlorine radical chemistry in the Eyjafjallajökull volcanic plume using observed depletions in non-methane hydrocarbons. *Geophysical Research Letters*, 38, L13801, doi: 10.1029/2011GL047571
- 29) Boichu M, Oppenheimer C, Roberts T, Kyle P, (2011): On bromine, nitrogen oxides and ozone depletion in the tropospheric plume of Erebus volcano (Antarctica). *Atmospheric Environment*, 45, 3856-3866
- 30) Jourdain L, Roberts TJ, Pirre M, Josse B, (2016): Modeling the reactive halogen plume from Ambrym and its impact on the troposphere with the CCATT-BRAMS mesoscale model. *Atmos. Chem. Phys.*, 16, 12099-12125
- 31) Roberts T, Martin RS, Jourdain L (2014): Reactive bromine chemistry in Mount Etna's volcanic plume: the influence of total Br, high temperature processing, aerosol loading and plume-air mixing. *Atmos. Chem. Phys.*, 14, 11201-11219
- 32) Roberts T, (2017): Halogen chemistry in volcanic plumes (invited). *Geophysical Research Abstracts*, 19, EGU2017, 7578
- 33) Sherwen T, Evans MJ, Carpenter LJ, Schmidt JA, Mickley LJ, (2017): Halogen chemistry reduces tropospheric O₃ radiative forcing. *Atmos. Chem. Phys.*, 17, 1557-1569
- 34) Kinnison D, Saiz-Lopez A, Lamarque J-F, Ordóñez C, Fernandez R, Tilmes S., (2013): Tropospheric chemistry and climate impacts of VSL halogens: pre-industrial to present day. *Geophysical Research Abstracts*, 15, EGU2013, 5896
- 35) Geng L, Murray LT, Mickley LJ, Lin P, Fu Q, Schauer AJ, Alexander B, (2017): Isotopic evidence of multiple controls on atmospheric oxidants over climate transitions. *Nature*, 546, 133-136
- 36) Thompson et al. (2017): Bromine atom production and chain propagation during springtime Arctic ozone depletion events in Barrow, Alaska. *Atmos. Chem. Phys.*, 17, 3401-3421
- 37) Wren SN, et al, (2013): Photochemical chlorine and bromine activation from artificial snow. *Atmos. Chem. Phys.*, 13, 9789-9800
- 38) Simpson WR, Peterson PK, Frieß U, Sihler H, Lampel J, Platt U, Moore C, Pratt K, et al, (2017). Horizontal and vertical structure of reactive bromine events probed by bromine monoxide MAX-DOAS spectroscopy. *Atmos. Chem. Phys. Discuss.*, doi: 10.5194/acp-2017-187
- 39) Custard KD, Thompson CR, Pratt KA, Shepson PB, Liao J, Huey LG, Oelando J, Weinheimer AJ, Apel E, Hall SR, Flocke F, Mauldin L, Hornbrook RS, Pöhler D, General S, Ziekle J, Simpson WR; Platt U, Fried A, Weibring P, Sive BC, Ullmann K, Cantrell C, (2015): The NO_x dependence of bromine chemistry in the arctic atmospheric boundary layer. *Atmos. Chem. Phys.*, 15, 10799-10809
- 40) Custard KD, Pratt KA, Wang S, Shepson PB, (2016): Constraints on Arctic atmospheric chloric production through measurements and simulations of Cl₂ and ClO. *Environ. Sci. Technol.*, 2016, 50(22), 12394-12400
- 41) Custard KD, Raso ARW, Shepson PB, Staebler RM, Pratt KA, (2017): Production and release of molecular Bromine and Chlorine from the arctic snowpack. *ACS Earth and Space Chemistry*, 1(3), 142-151
- 42) Nasse J-M, Frieß U, Pöhler D, Schmitt S, Weller R, Schaefer T, Platt U, (2017). Unexpected autumnal halogen activity in the lower atmosphere at Neumayer III/Antarctica. *Geophysical Research Abstracts*, 19, EGU2017, 3888
- 43) Oum KW, Lakin MJ, Finlayson-Pitts BJ, (1998): Bromine activation in the troposphere by the dark reaction of O₃. *Geophysical Research Letters*, 25(21), 3923-3926
- 44) Ferrari R, Jansen MF, Adkins JF, Burke A, Steward AL, Thompson AF, (2014): Antarktis sea ice control on ocean circulation in present and glacial climates. *PNAS*, 111(24), 8753-8758.
- 45) Kebede MA, Bish DL, Losovjy J, Engelhard MH, Raff JD, (2016): The role of iron-bearing minerals in NO₂ to HONO conversion on soil surfaces. *Environmental Science and Technology*, 50, 8649-8660