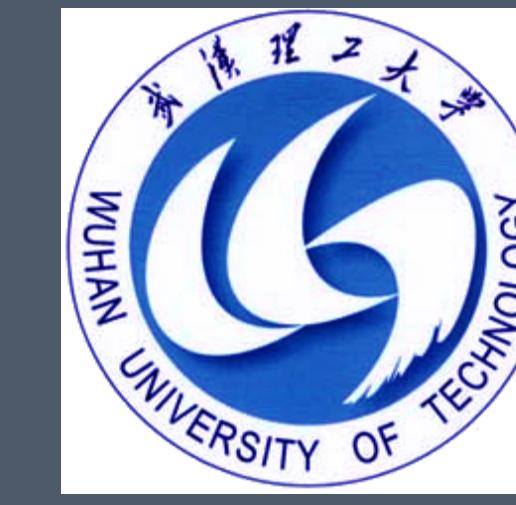


# 250 years of research revealed Gaia's climate control – there is sufficient knowledge now for technical CE tests inspired from Nature's "CE method"



résumé



**1758 / 1804:** The Russian Count Bestuschew-Rjumin used the FeCl<sub>3</sub> photolysis to prepare his Nerve Tincture<sup>1</sup>). A. F. Gehlen (1804) investigated the <sup>35</sup>Cl generating photolysis step of the tinctures preparation: sunshine photon +Fe(III)Cl<sub>3</sub> → Fe(II)Cl<sub>2</sub> + <sup>35</sup>Cl<sup>2</sup>.

**1990 / 2017:** J.H. Martin formulated "The Iron Hypothesis": the glacials are induced by phytoplankton fertilized iron dust<sup>3</sup>). In the Antarctic, the ice cores reveal that during glacial ages there was high dust, low CO<sub>2</sub>, and low methane levels and that during warm interglacials there was low dust, high CO<sub>2</sub> and high methane levels<sup>13</sup>). Dust concentrations during glacials have been 50 to 70 times higher than during warm interglacials<sup>14</sup>). Glacial outwash plains are potent sources of iron-containing sub-micron dust<sup>15-16</sup>). The solubilisation of atmospheric Fe mineral aerosol particles by HCl is activated by sunshine<sup>17</sup>).

**2004:** F. D. Oeste proposed tropospheric methane depletion by atomic <sup>35</sup>Cl generated from Fe(III) containing dust, sunshine and chloride<sup>4</sup>).

**2003 / 2012:** Additional to sediments, the solidified ocean crust (up to 2 km below the ocean bottom) acts as flexible CO<sub>2</sub> carbon dump: this fissured convective flow system has recharge and discharge zones<sup>5</sup>). At an ocean crust growth rate of about 24 km<sup>3</sup>/yr = 7 x 10<sup>10</sup> t/yr<sup>6</sup>) it fixes recent HCO<sub>3</sub><sup>-</sup> and CO<sub>3</sub><sup>2-</sup> as calcite by up to 1.6 weight % = 1.1 x 10<sup>9</sup> t/yr or 0.48 x 10<sup>9</sup> t CO<sub>2</sub>/yr<sup>7</sup>). The elevated Cretaceous CO<sub>2</sub> level of 1130 ppm<sup>8</sup>) increased the ocean crust calcite content to 5.5 %<sup>7</sup>).

**2012 / 2017:** Direct vegetation climate control acts by CO<sub>2</sub> assimilation and HCO<sub>3</sub><sup>-</sup> and calcite generation by accelerated weathering<sup>9</sup>)<sup>10</sup>)<sup>11</sup>): Roots and mycorrhizal fungi excrete CO<sub>2</sub>, organic acids, complexing agents, and generate fissures by mechanical opening. Roots transform CO<sub>2</sub> and rock to hydrogen carbonate and soil at velocities many times faster than abiotic weathering. Roots increase CO<sub>2</sub> level of soil air up to 100 times that of atmosphere. Foliar fertilization by soluble iron salts of mineral aerosol support and accelerates vegetation weathering especially of those plants growing on alkaline rock and soil types<sup>12</sup>).

**2004 / 2016:** Confirmation of activated photochemical production of <sup>35</sup>Cl atoms and/or <sup>35</sup>Br atoms from iron(III) and solid, liquid or aerosol halogenid<sup>18</sup>)<sup>19</sup>)<sup>20</sup>) or from iron containing aerosol in the presence of gaseous HCl<sup>21</sup>)<sup>22</sup>). Iron oxides and dissolved iron in aerosols and tropospheric H<sub>2</sub>O<sub>2</sub> induce atmospheric Fenton reactions and produce atomic <sup>35</sup>Cl and <sup>35</sup>OH radicals in the troposphere even in the dark<sup>23</sup>)<sup>24</sup>)<sup>25</sup>).

**2016:** Confirmation of the photochemical generation of CH<sub>4</sub> depleting <sup>35</sup>OH radicals by iron(II) aerosol and NO<sub>2</sub> 45)

**2017:** A technical variant mimicking the natural "Iron Salt Aerosols (ISA)" climate control method has been proposed<sup>12</sup>). 12 or more cooling effects are described, among them CH<sub>4</sub> and CO<sub>2</sub> atmospheric removal.

conclusions

## Does current research ignore important climate regulation principles?

### Ignored: halogen activation by iron photolysis (Figure 1)

**1989:** Revelation of the O<sub>3</sub>-induced atomic <sup>35</sup>Cl generation from gaseous HCl by Fe<sub>2</sub>O<sub>3</sub> aerosol photolysis. This process shall contribute to the chemistry of Antarctic O<sub>3</sub> depletion<sup>26</sup>).

**2009 / 2017:** Discussions about the cause of atomic <sup>35</sup>Cl and <sup>35</sup>Br in volcanic eruption plumes<sup>27</sup>)<sup>28</sup>)<sup>29</sup>)<sup>30</sup>)<sup>31</sup>)<sup>32</sup>).

**2013 / 2017:** Models about the influence of mankind induced halogen emissions on tropospheric O<sub>3</sub> content<sup>33</sup>)<sup>34</sup>).

**2017:** Discussion about the increased tropospheric halogen chemistry during glacial climate periods<sup>35</sup>).

**1998 / 2017:** Discussion about possibilities of the generation of atomic <sup>35</sup>Cl and <sup>35</sup>Br from arctic snow and sea ice<sup>36</sup>)<sup>37</sup>)<sup>38</sup>)<sup>39</sup>)<sup>40</sup>)<sup>41</sup>)<sup>42</sup>)<sup>43</sup>).

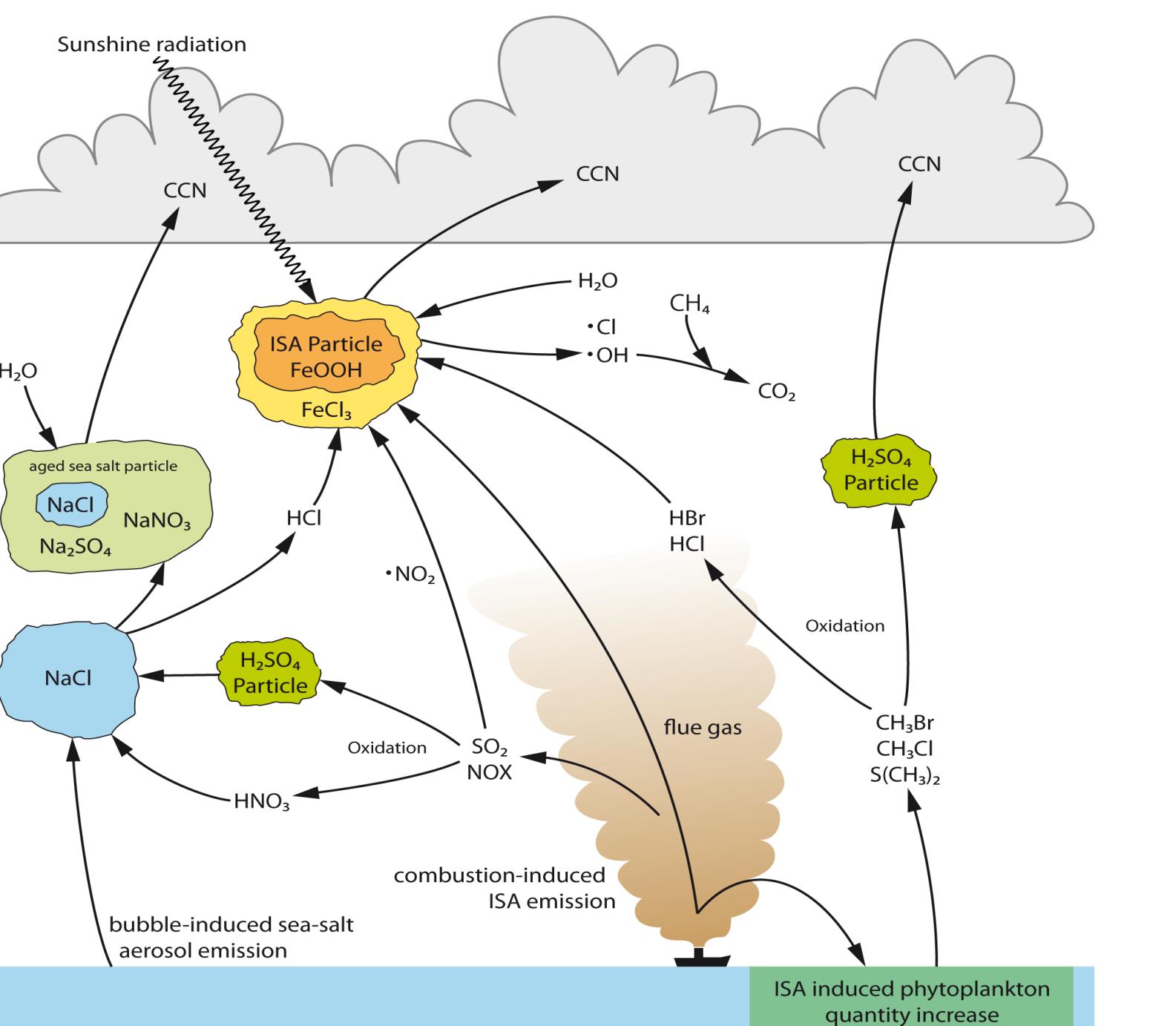
### Ignored: carbon cycle direction

#### atmosphere → ocean → sediment / crust; not vice versa (Scheme 1):

**2014:** Upwelling water in the Antarctica neighboring glacial Southern Ocean should have lost lesser CO<sub>2</sub> to the atmosphere than recently<sup>44</sup>).

## FIGURE 1

### Tropospheric CH<sub>4</sub> depletion induced by Fe(III) photolysis



Scheme 1

### Driving-forces directing the carbon cycle from atmosphere into ocean and from ocean into sediments and crust

**Troposphere:** Oxidation of CH<sub>4</sub>, CO, COS and VOC to CO<sub>2</sub> and soot hydrophilization  
→ → →

**Continent/ocean:** 1) CO<sub>2</sub> absorption  
2) CO<sub>2</sub> hydrolysis to HCO<sub>3</sub><sup>-</sup>,  
3) CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> metabolism to organic and carbonate C  
4) sedimentation of organic and carbonate C  
→ → →

**Sediment/oceanic crust:** Carbonate and CH<sub>4</sub> hydrate precipitation  
→ → →

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## BIBLIOGRAPHY

- 1) Weitbrecht JJ. (1780): Nachricht von der Zubereitung der gelben und weißen Bestuschewskens Nerven-Tinktur. St. Petersburgisches Journal, 10, 184-189
- 2) Gehlen AF. (1804): Neues allgemeines Journal der Chemie, 3, 566; aus Landgrebe DG, (1834): Über die chemischen und physiologischen Wirkungen des Lichtes. Druck und Verlag von N.G. Elwert, Marburg
- 3) Martin JH. (1990): Glacial-interglacial CO<sub>2</sub> change: the iron hypothesis. Paleoceanography, 5, 1-13
- 4) Oeste FD. (2004): Climate cooling by interaction of natural (A) or artificial (B) loess dust with tropospheric methane. GeoLeipzig 2004, Deutsche Geologische Gesellschaft, Hannover 2004, p. 344, ISBN 3-932537-06-8
- 5) Smith-Duque C. (2009): Hydrothermal alteration of upper oceanic crust formed at fast spreading rates. Thesis, University of Southampton, Faculty of Engineering, Science & Mathematics, School of Ocean & Earth sciences
- 6) Staudigel H (2003): 3.15 Hydrothermal alteration processes in the ocean crust. Treatise on Geochemistry, Volume 3, Elsevier, 511-535, ISBN 0-08-044338-9
- 7) Rausch S. (2012): Carbonate veins as recorders of seawater evolution, CO<sub>2</sub> uptake by ocean crust, and seawater-crust interaction during low temperature alteration. Thesis, Universität Bremen, Fachbereich Geowissenschaften
- 8) Fletcher BJ, Brennand CW, Anderson CJ, Berner RA, Beerling DJ. (2008): Atmospheric carbon dioxide linked with Mesozoic early Cenozoic climate change. Nature Geoscience, 1, 43-48
- 9) Taylor LL, Banwart SA, Valdes PJ, Leake RL, Beerling DJ, (2012): Evaluation the effects of terrestrial ecosystems, climate and carbon dioxide on weathering over geological time: a global-scale process-based approach. Phil. Trans. R. Soc. B, 367, 565-582
- 10) Hartmann J, West AJ, Renforth P, Köhler P, De la Rocha, CL, et al. (2013): Enhanced chemical weathering as a geoengineering strategy to reduce atmospheric carbon dioxide, supply nutrients and mitigate ocean acidification. Reviews of Geophysics, 51, 113-150
- 11) Kronfeld J, Moinester M, Carmi I, Godfrey-smith D, (2015): Sequestration of inorganic carbon via Forestation. Presented at Climate Engineering Research SymposiumJuly 2015, Berlin, Germany
- 12) Oeste FD, de Richter R, Ming T, Caillol S. (2017): Climate engineering by mimicking natural dust climate control: the iron salt method. Earth Syst. Dynam. 8, 1-54
- 13) Masson-Delmotte V, Stenni B, Pol K, Braconnot P, Cattani O, Falourd S, Kageyama M, Jouzel J, Landais A, Minster B, Barnola JM, Chappellaz J, Krinner G, Johnson S, Röhlisberger R, Hansen J, Mikalajewicz U, Otto-Biesner B, (2010): EPICA Dome C record of glacial and interglacial intensities. Quaternary Science Reviews, 29, 113-128
- 14) Liu Y, Shi G, Xie Y, (2013): Impact of dust aerosol on glacial-interglacial climate. Advances in Atmospheric Sciences, 30(6), 1725-1731
- 15) Dagsson-Waldhauserova P, Arnalds O, Olafsson H, Magnusdottir A, (2017): High proportions of sub-micron particulate matter in Icelandic dust storms in 2015. Geophysical Research Abstracts, 19, presented at EGU2017, 8762-1
- 16) Kubin A, Schepanski K, Heinold B, Tegen I, (2017): Sensitivity studies on glaciogenic dust mobilization in Greenland with a mesoscale model. Geophysical Research Abstracts, 19, presented at EGU2017, 3207
- 17) Rubasinghege G, Lentz RW, Scherer MM, Grassian VH, (2010): Simulated atmospheric processing of iron oxyhydroxide minerals at low pH: roles of particle size and acid anion in iron dissolution. PNAS, 107(15), 6628-6633
- 18) Lim M, Chiang K, Amal R, (2006): Photochemical synthesis of chlorine gas from iron(III) and chloride solution. Journal of Photochemistry & Photobiology A: Chemistry, 183(1-2), 126-132
- 19) Wittmer J, Bleicher S, Ofner J, Zetsch C, (2015a): Iron(III)-induced activation of chloride from artificial sea-salt aerosol. Environmental Chemistry, 12(4), 461-475
- 20) Wittmer J, (2015b): Photochemical activation of chlorine and bromine from iron-doped saline media. Dissertation at the Biological, Chemical and Geosciences Faculty of the University of Bayreuth, Germany
- 21) Wittmer J & Zetsch C, (2016): Photochemical activation of chlorine by iron-oxide aerosol. Journal of Atmospheric Chemistry, 10874, 1-18
- 22) Vione D, Maurino V, Minero C, Pelizzetti E, Harrison MAJ, Olariu R-J, Arsene C, (2006): Photochemical reactions in the tropospheric aqueous phase and on particulate matter. Chem. Soc. Rev., 35, 441-453
- 23) Luna AJ, Nascimento CAO, Chiavone-Vilho O, (2006): Photodecomposition of hydrogen peroxide in highly saline aqueous medium. Brazilian Journal of Chemical Engineering, 23(3), 341-349
- 24) Machalek A, Moraes JEF, Okano LT, Silvério CA, Quina FH, (2009): Photolysis of ferric ions in the presence of sulfate or chloride ions: implications for the photo-Fenton process. Photocem. Photobiol. Sci., 8, 985-991
- 25) Machalek A, Quina FH, Gozzi F, Silva VO, Friedrich LC, Moraes JEF, (2012): Chapter 11 – Fundamental mechanistic studies of the photo-Fenton reaction for the degradation of organic pollutants. In the book titled "Organic pollutants ten years after the Stockholm Convention – Environmental and Analytical update" edited by Puzyń T and Mostrag-Szlichtyng A. ISBN 978-39353-307-917-2
- 26) Behnke W & Zetsch C, (1989): Photochemical formation of Cl atoms from NaCl and HCl in the presence of ozone and aerosol. Our changing Atmosphere, Proceedings of the 28th Liège International Astrophysical Colloquium. Edited by P.J. Crutzen, J.-C. Gerard and R.Zander, 493-499
- 27) von Glasow R, Bobrowski N, Kern C, (2009): The effects of volcanic eruptions on atmospheric chemistry. Chemical Geology, 1-4, 131-142
- 28) Baker AK, Rauthal-Schöch A, Schuck TJ, et al, (2011): Investigation of chlorine radical chemistry in the Þórafljóður volcanic plume using observed depletions in non-methane hydrocarbons. Geophysical Research Letters, 38, L13801, doi: 10.1029/2011GL047571
- 29) Boichu M, Oppenheimer C, Roberts T, Kyle P, (2011): On bromine, nitrogen oxides and ozone depletion in the tropospheric plume of Erebus volcano (Antarctica). Atmospheric Environment, 45, 3856-3866
- 30) Jourdain L, Roberts TJ, Pirre M, Josse B, (2016): Modeling the reactive halogen plume from Ambrym and its impact on the troposphere with the CCATT-BRAMS mesoscale model. Atmos. Chem. Phys., 16, 12099-12125
- 31) Roberts T, Martin RS, Jourdain L, (2014): Reactive bromine chemistry in Mount Etna's volcanic plume: the influence of total Br, high temperature processing, aerosol loading and plume-air mixing. Atmos. Chem. Phys., 14, 11201-11219
- 32) Roberts T, (2017): Halogen chemistry in volcanic plumes (invited). Geophysical Research Abstracts, 19, EGU2017, 7578
- 33) Sherwin T, Evans MJ, Carpenter LJ, Schmidt JA, Mickley LJ, (2017): Halogen chemistry reduces tropospheric O<sub>3</sub> radiative forcing. Atmos. Chem. Phys., 17, 1557-1569
- 34) Kinnison D, Saiz-Lopez A, Lamarque J-F, Ordonez C, Fernandez R, Tilmes S., (2013): Tropospheric chemistry and climate impacts of VSL halogens: pre-industrial to present day. Geophysical Research Abstracts, 15, EGU2013, 5896
- 35) Genz L., Murray LT, Mickley LJ, Lin P, Fu Q, Schaefer AJ, Alexander B, (2017): Isotopic evidence of multiple controls on atmospheric oxidants over climate transitions. Nature, 546, 133-136
- 36) Thompson et al. (2017): Bromine atom production and chain propagation during springtime Arctic ozone depletion events in Barrow, Alaska. Atmos. Chem. Phys., 17, 3401-3421
- 37) Wren SN, et al., (2013): Photochemical chlorine and bromine activation from artificial snow. Atmos. Chem. Phys., 13, 9789-9800
- 38) Simpson WR, Peterson PK, Frieß U, Sihsler H, Lampel J, Platt U, Moore C, Pratt K, et al, (2017): Horizontal and vertical structure of reactive bromine events probed by bromine monoxide MAX-DOAS spectroscopy. Atmos. Chem. Phys. Discuss., doi: 10.5194/acp-2017-187
- 39) Custard KD, Thompson CR, Pratt KA, Shepson PB, Liao J, Huey LG, Oelando JJ, Weinheimer AJ, Apel E, Hall SR, Flocke F, Mauldin L, Hornbrook RS, Pöhler D, General S, Zielke J, Simpson WR; Platt U, Fried A, Weibring P, Sive BC, Ullmann K, Cantrell C, (2015): The NO<sub>x</sub> dependence of bromine chemistry in the arctic atmospheric boundary layer. Atmos. Chem. Phys. 15, 10799-10809
- 40) Custard KD, Pratt KA, Wang S, Shepson PB, (2016): Constraints on Arctic atmospheric chlorine production through measurements and simulations of Cl<sub>2</sub> and ClO. Environ. Sci. Technol., 2016, 50(22), 12394-12400
- 41) Custard KD, Raso ARW, Shepson PB, Staebler RM, Pratt KA, (2017): Production and release of molecular Bromine and Chlorine from the arctic snowpack. ACS Earth and Space Chemistry, 1(3), 142-151
- 42) Nasse J-M, Frieß U, Pöhler D, Schmitt S, Weller R, Schaefer J, Platt U, (2017): Unexpected autumnal halogen activity in the lower atmosphere at Neumayer III/Antarctica. Geophysical Research Abstracts, 19, EGU2017, 3888
- 43) Oum KW, Lakin MJ, Finlayson-Pitts BJ, (1998): Bromine activation in the troposphere by the dark reaction of O<sub>3</sub>. Geophysical Research Letters, 25(21), 3923-3926
- 44) Ferrari R, Jansen MF, Adkins JF, Burke A, Steward AL, Thompson AF, (2014): Antartikts sea ice control on ocean circulation in present and glacial climates. PNAS, 111(24), 8753-8758
- 45) Kebede MA, Bish DL, Losovoy Y, Engelhardt MH, Raff JD, (2016): The role of iron-bearing minerals in NO<sub>2</sub> to HONO conversion on soil surfaces. Environmental Science and Technology, 50, 8649-8660